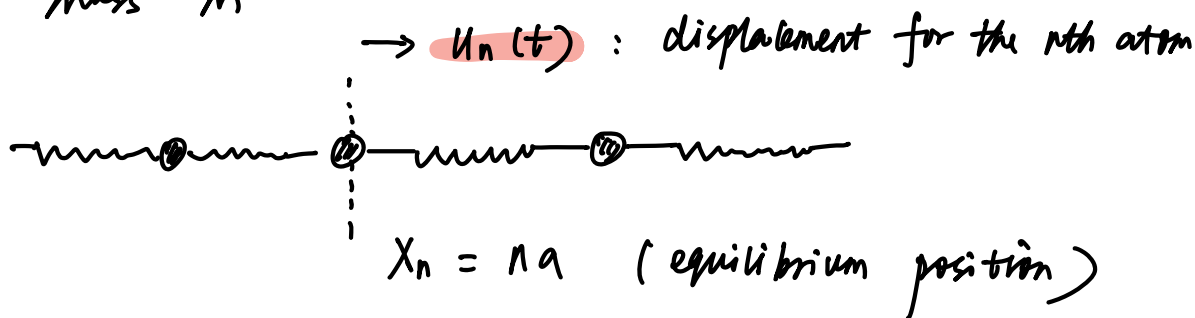


refer to prof. Hsin-Hau Lin

sound waves propagating in solids and fluids  
& We have biological receptors for it.

Approximate the 1D solids as an array of coupled atoms mass  $M$



$$u(x,t) = U_n(t)$$

$$\begin{aligned} M \ddot{U}_n &= -K(U_n - U_{n-1}) + K(U_{n+1} - U_n) \\ &= -K a \left. \frac{\partial u}{\partial x} \right|_{n-\frac{1}{2}} + K a \left. \frac{\partial u}{\partial x} \right|_{n+\frac{1}{2}} \\ &= a^2 K \left. \frac{\partial^2 u}{\partial x^2} \right|_n \end{aligned}$$

$$M \frac{\partial^2 u}{\partial t^2} = a^2 K \frac{\partial^2 u}{\partial x^2}$$

$$v = \sqrt{\frac{K}{M}} a$$



$$\omega_0 = \sqrt{\frac{K}{M}}$$

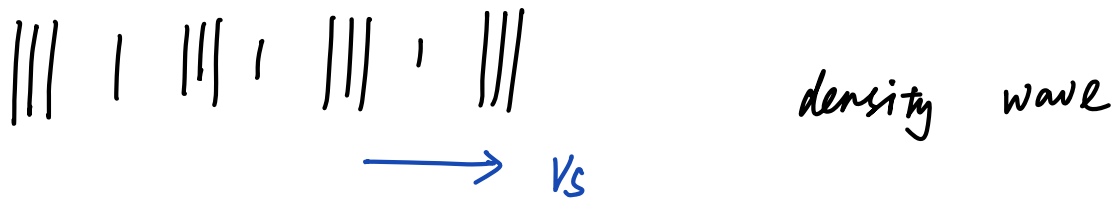
vibration

For stiff materials, the spring constant  $K$  is large  $\rightarrow$

the sound speed is fast. Note that  $v \sim 6000$  m/s in steel,  
much faster than  $v \sim 330$  m/s in air.

# Sound waves in an ideal gas

interesting observation



Q: what is the relevant parameters?

$P, \rho, T, m, \dots$

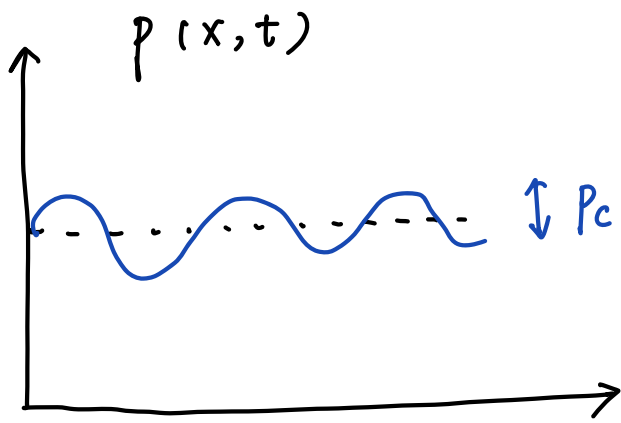
what is your intuitive guess?

thermodynamic & microscopic quantities

A rough picture for the sound wave helps:

- gas molecules move and change the density.
- change in density  $\rightarrow$  change in pressure
- pressure variations generate molecular motions.

The above picture suggests that we can identify  $P$  and  $\rho$  as the relevant parameters.



$$\begin{cases} p(x, t) = p_0 + p_c(x, t) \\ p(x, t) = p_0 + p_c(x, t) \end{cases}$$

( $p_0, p_0$  for equilibrium)

$$P = f(\rho)$$

It is reasonable to assume that  $p_c \ll p_0$   
and  $\rho_c \ll \rho_0$  in usual sound waves

$$\begin{aligned} p_0 + p_c &= f(\rho_0 + \rho_c) \\ &= f(\rho_0) + f'(\rho_0) \rho_c + \frac{1}{2} f''(\rho_0) \rho_c^2 + \dots \end{aligned}$$

keeping the lowest non-vanishing term

$$p_c(x, t) = k \rho_c(x, t)$$

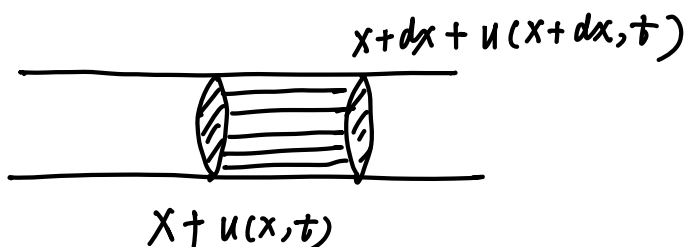
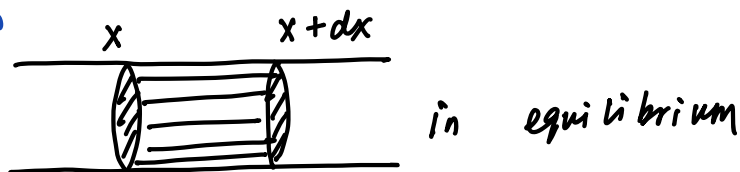
$$k = f'(\rho_0) = \left( \frac{dp}{d\rho} \right)_0$$

the generalized Hooke's law again:

$$p_c \propto \rho_c$$

the variation of the pressure  $\propto \rho$  of the density

demonstration



$u(x, t)$ :  
the displacement at  
position  $x$  and time  $t$

From mass conservation in the tiny segment: ( $x \rightarrow x + dx$ )

$$\rho_0 dx = \rho [x + dx + u(x + dx, t) - (x + u(x, t))]$$

$$= \rho [dx + \frac{\partial u}{\partial x} dx]$$

$$\rho_0 = \rho (1 + \frac{\partial u}{\partial x}) = (\rho_0 + \rho_c) (1 + \frac{\partial u}{\partial x})$$

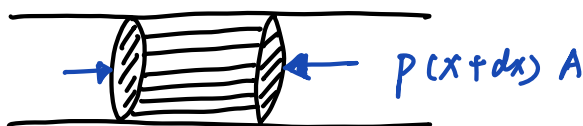
$$= \rho_0 + \rho_c + \rho_0 \frac{\partial u}{\partial x} + \cancel{\rho_c \frac{\partial u}{\partial x}} = 0$$

(smallness)

$$( \rho_c / \rho_0 \ll 1, \partial u / \partial x \ll 1 )$$

Finally, we relate the density variation  $\rho_c(x, t)$  to the spatial derivative of the scalar field  $u(x, t)$

$P(x)A$



$$\rho_c(x, t) = -\rho_0 \frac{\partial u}{\partial x}$$

$$[ -P(x+dx) + P(x) ] A = \cancel{\rho_0 A dx} \frac{\partial^2 u}{\partial t^2} \quad (\text{Newton 2nd law})$$

$$= - \frac{\partial P}{\partial x} dx = - \frac{\partial \rho_c}{\partial x} dx = -k \frac{\partial \rho_c}{\partial x} dx = k \rho_0 \frac{\partial^2 u}{\partial x^2} dx$$

$$\cancel{k \rho_0} \frac{\partial^2 u}{\partial x^2} dx = \cancel{\rho_0 dx} \frac{\partial^2 u}{\partial t^2} \Rightarrow \frac{\partial^2 u}{\partial t^2} = k \frac{\partial^2 u}{\partial x^2}$$

wave equation  $V^2 = k = \left( \frac{dP}{d\rho} \right)_0$

similar to the derivation in the elastic wave

$$v^2 = K = \left( \frac{dP}{d\rho} \right)_0$$

dimension analysis:

$$P V_{\text{vol}} = N k_B T$$

$$P = \text{energy} / V_{\text{vol}}$$

$$\rho = \frac{M}{V_{\text{vol}}} = \frac{\text{energy}}{v^2 V_{\text{vol}}}$$

because the oscillation of the sound wave is quite fast, it can be viewed as adiabatic process (no heat in or out during oscillation)

$$P V_{\text{vol}}^\gamma = \text{const}$$

$$P = \text{const } \rho^\gamma = C \rho^\gamma$$

$$\left( \frac{dP}{d\rho} \right)_0 = \gamma C \rho^{\gamma-1} = \frac{\gamma P}{\rho}$$

$$v = \sqrt{\frac{\gamma P}{\rho}}$$

What about the process handled by  
the isothermal one ?

$$P V_{\text{vol}} = \text{const}$$

$$P = c \rho$$

$$V = \sqrt{\frac{P}{\rho}}$$

You will make the same mistake as  
Newton did! (more than 10% molecules  
so proud of you <sup>^o</sup> not involved?)

$$V = \sqrt{\frac{\gamma P}{\rho}}$$

due to  $P V_{\text{vol}} = N k_B T$

$$P = \frac{N}{V_{\text{vol}}} k_B T$$

$$= \frac{N m}{m V_{\text{vol}}} = \frac{P V_{\text{vol}}}{m V_{\text{vol}}} k_B T$$

$$= \frac{P}{m} k_B T$$

(  $m$  : mass of a molecule )

$$\therefore V = \sqrt{\frac{3k_B T}{m}}$$

$$V_{\text{rms}} = \sqrt{\overline{V^2}} \quad (\text{root mean square})$$

$$= \sqrt{\frac{3k_B T}{m}}$$

Newton's mechanics works again!  
and explains the propagation of sound waves

$$k_B : 1.38 \times 10^{-23} \text{ J/K}$$

$$(\text{dry air}) \quad T : 300 \text{ K (room temperature)}$$

$$28.9647 \text{ g/mol} \quad m : \text{N}_2 (78 \text{ vol\%}), \text{O}_2 (21 \text{ vol\%})$$

The molar mass of a substance is defined as the mass of 1 mol of that substance expressed in grams per mole, and is equal to the mass of  $6.022 \times 10^{23}$  atoms, molecules

$$m_u = 1 \text{ g/mol}$$

$$V = \sqrt{\frac{1.4 \times 1.38 \times 10^{-23} \times 300}{28.9647 \times 10^{-3} / 6.02 \times 10^{23}}} \approx 347 \text{ m/s}$$